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1. Model 1 shows data collected from two experiments where electric current was run through molten alloy containing the desired metal. a. Will the desired metal be collected at the anode or the cathode of the cell? The desired metal will be collected at the cathode. b. Write the half reaction that occurs as the metals are collected on the electrode.

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Model 1 - Gases in a Nonflexible Container Experiment A ... 186 POGIL™ Activities for ... Which gas variable (P internal, V, T or n) is most closely related to the length of the arrows in Model 1? Temperature is related to average kinetic energy, and thus to the length of the arrows. 4. Complete the following table for the two experiments ...

~~Model 1 Gases in a Nonflexible Container Experiment A ...~~
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4 POGIL™ Activities for High School Chemistry Model 2 - Gases in a Flexible Container Experiment C (Adding more gas) C1 C3C2
Volume = 1 unit Volume = 2 units Volume = 3 units External pressure = 1 atm External pressure = 1 atm External pressure = 1 atm

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4 POGIL™ Activities for AP* Biology Model 2 ... How is the dependent variable in Model 2 different from that in Model 1? The dependent variable in Model 1 was the number of families that went extinct versus the dependent variable in Model 2 is how many families are living. 13. What is the overall trend shown in the graph in Model 2?

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Chem 116 POGIL Worksheet - Week 1 Gas Laws - Part 1 Why? Of the three principal states of matter (gas, liquid, solid), gases show behavior that is most easily connected to molecular motion. The observed behavior of gases, embodied in the empirical gas laws, leads to a series of equations that can be summarized by a single equation of state, called

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method for measuring the rate of gas production from the aquatic plant in Model 1. 3. With your group, create a list of environmental factors that may affect the rate of gas production in the aquatic plant in Model 1. These factors could become variables in an experiment.

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3. In Model 1, the length of the arrows represents the average kinetic energy of the molecules in that sample. Which gas variable (P internal, V, T or n) is most closely related to the length of the arrows in Model 1? 4. Complete the following table for the two experiments in Model 1. Experiment A Experiment B Independent Variable Dependent ...

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4 POGIL™ Activities for High School Chemistry Model 2 - Gases in a Flexible Container Experiment C (Adding more gas) C1 C3C2
Volume = 1 unit Volume = 2 units Volume = 3 units External pressure = 1 atm External pressure = 1 atm External pressure =

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1 atm

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~~KEY Mass Extinctions~~

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~~Why? Model 1 — Photosynthesis in an Aquatic Plant~~

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