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cientiic Inc ll iht eered Write the
actual mass here 0.98 x (correct
answer) 0.95 x (correct answer)
0.90 x (correct answer) 1.02 x

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(correct answer) 1.05 x (correct answer) 1.10 x (correct answer)
Correct answer: (actual mass divided by 63.546 times
6.022e23) 4 3 2 Figure 1.
Procedure 1.

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actual mass here 0.98 x (correct
answer) 0.95 x (correct answer)
0.90 x (correct answer) 1.02 x
(correct answer) 1.05 x (correct
answer) 1.10 x (correct answer)
Correct answer: (actual mass

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Moles Lab Activity 3: Compounds

—Water, Compounds Sodium

Chloride, Compounds Chalk

Compounds—The Fictitious

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Compound “Cambium” Moles Lab
Activity 4: Solutions—Aqueous
Copper (II) Sulfate Pent hydrate,
Solutions—Alum Moles Lab
Activity 5: Synthesis of an Oxide
of Copper

Moles Lab Activities

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The fastest way to obtain a
relative mass of beans would be

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to count the beans. The fastest way to obtain a mole of beans would be to weigh them. (At least in principle. The mass of a mole of beans would be incredibly large- on the order of 10^{22} g.)
Part III. All atomic masses agree with the relative masses to three

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significant figures.

Laboratory Activity 1: Teacher
Notes Continued
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Answers

Have a conceptual understanding of the mole as the method of "counting" items and finding the mass of items that can't be seen.

2. Be able to calculate the number of items (molecules,

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atoms, ions, and formula units) if given the number of moles. 3. Be able to calculate the number of moles if given the number of items.

#20 Introduction to the Mole -
Terrific Science

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continue 5 216 linn cientiic inc all
ights esered target gas law lab
worksheet many chemical
reactions involve gaseous
reactants or products the
volumes of gases that react or are
produced depend entirely on the

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