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present and explain, at
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• In principle, solve Schrödinger Equation •

Not possible for many-electron atoms or molecules due to many-body problem

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The foundation of
quantum mechanics
and quantum
chemistry is the wave
model, in which the
atom is a small, dense,
positively charged
nucleus surrounded by
electrons. The wave
model is derived from

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the wave function , a set of possible equations derived from the time evolution of the Schrödinger equation which is applied to the wavelike probability distribution of subatomic particles.

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Postulates of quantum mechanics (short) (PI)

State is given by a wave function $U'(r, t)$.

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2 Quantum Mechanics
Made Simple
communication,
quantum cryptography,
and quantum
computing. It is seen
that the richness of
quantum physics will
greatly affect the future
generation
technologies in many
aspects. 1.2 Quantum
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The development of
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a great intellectual

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The book is largely self-contained and requires no prerequisite other than a solid undergraduate physical chemistry course; however, some exposure to quantum chemistry will enhance the student's appreciation of the material.

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