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There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas. These three states are quite different. The main difference is in their particles.

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Chapter 13: States of Matter -
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Section 13.4 Changes of State

The relationship among the solid, liquid, and vapor states (or phases) of a substance in a sealed container are best represented in a single graph called a phase diagram. Phase diagram- gives the temperature and pressure at which a substance exists as solid, liquid, or gas (vapor) 70.

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Vocabulary from Chapter 13

"Phases of Matter" for Mr. Testa's Honors Chemistry class. Still a work in progress. Currently only contains some gas laws and properties.

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Chapter 13 States of Matter 137

SECTION 13.1 THE NATURE OF

GASES (pages 385–389) This

section introduces the kinetic

theory and describes how it

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States of Matter. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386)
1.

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