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(b) The atomic weights of the elements ordinarily are not integers because: (1) the atomic masses of the atoms generally are not integers (except for. 12. C), and (2) the atomic weight is taken as the weighted average of the atomic masses of an atom's naturally occurring isotopes.

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\* Clear and concise discussions This text has received many accolades for its ability to clearly and concisely convey materials science and engineering concepts at an appropriate level to ensure student understanding. For examples see chapters 3, 4, 5 and 9. \* Mechanical property coverage The Sixth ...

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qemt Material Science & Engineering NUST Preview text CHAPTER 2 ATOMIC STRUCTURE AND INTERATOMIC BONDING PROBLEM SOLUTIONS Fundamental Concepts Electrons in Atoms 2.1 Cite the difference between atomic mass and atomic weight.

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